Relative permeability prediction considering complex pore geometry and wetting characteristics in carbonate reservoirs

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Abstract: The measurement of relative permeability in carbonate rocks is very uncertain because of the complex pore system. Several equations for the relative permeability have been previously developed. However, most equations assume a single pore system that cannot be applied accurately. This study presents a relative permeability estimation method that considers capillary pressure, contact angle, pore size distribution, and residual oil saturation with respect to the heterogeneous pore network. As results, the capillary pressure was observed to have different tendencies for macropore and micropore. Also, the pore size distribution index of the macropore system was greater than the micropore system. For the macropore, water has a higher relative permeability than oil; a small reduction in oil saturation can easily flow water. In contrast, the micropore tends to follow the conventional relative permeability curve. Therefore, stable water displacement was not assured, leading to an early breakthrough for the heterogeneous carbonate rocks. [Received: July 11, 2016; Accepted: January 2, 2017]

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1 Introduction

In carbonate rocks, secondary pores such as dissolved pores, vugs, cavities, and fractures are well developed. Thus, carbonate rocks have heterogeneous pore systems caused by carbonate diagenesis. Most carbonate reservoirs have oil-wet characteristics, affecting not only fluid flow in the heterogeneous pore system but also oil production. In carbonate reservoirs, relative permeability measurement is extremely difficult because of the heterogeneous pore networks and wettability characteristics. In addition, relative permeability measurement is time consuming and has exhibited significant variation across experimental results, even for samples taken from the same formation. A number of correlations have been previously developed to explain multiphase flow in porous mediums; however, they cannot accurately predict the relative permeability of carbonates because they generally assume a homogeneous pore system and single pore size. In this study, we measured the pore size distribution, capillary pressure, and contact angle in a heterogeneous carbonate rock with wide-ranging pore sizes. Relative permeability curves were generated for the micropore and macropore using capillary pressure data, respectively. In carbonate rocks, there are not experimental identifications of the capillary pressure considering both micropore and macropore simultaneously. This study notably observed relative permeability characteristics by measuring capillary pressure for the different pore sizes. The results from this study can be used to explain the complex flow behaviour in heterogeneous carbonate reservoirs.

2 Background

2.1 Carbonate reservoir characteristics

More than 60% of the world's oil reservoirs are made of carbonate rock, and more than 80% of the carbonate rock has oil-wet characteristics. Carbonate rock that does not make contact with oil tends to have water-wet characteristics. Rao (1996) purported that the wettability of carbonate rock is changed to oil-wet by the chemical reaction between the rock surface and the oil surface activation components; when oil pressure exceeds critical capillary pressure, the thin layer of connate water on the rock surface is destroyed. Hamouda and Karam (2006) considered the altered wettability of carbonate rock. Due to the loss of hydrogen ion, the carboxyl group of oil presents a negative electric charge and adheres to the positively charged rock surface. For the oil-wet rock, water flows more easily than oil through the pore networks that oil recovery can be less than 10% (Montaron, 2005). Water flooding simulations show that only limited amounts of oil can be recovered from oil-wet layers because the water tends to flow primarily through the water-wet layers in carbonate reservoirs (Schlumberger, 2007). The heterogeneous pore system and difficulties in fluid flow prediction require an enhanced understanding of fundamental carbonate reservoir characteristics.

2.2 Pore size distribution

Heterogeneous pore systems comprising dissolved pores, vugs, cavities, and fractures significantly affect permeability. For this reason, it is important to analyse reservoir properties with respect to pore size. Hwang (2010) proposed a more precise prediction model for drainage capillary pressure, which considers micropore and macropore size effects in heterogeneous carbonate rock. He confirmed that this approach was more reliable than existing capillary pressure models that assume a homogeneous pore system. Accordingly, pore size analysis is essential for heterogeneous carbonate reservoirs. Several pore size measurement methods exist, including the gas adsorption, mercury injection, micro-tomography, and electron microscopy methods. Among these methods, gas adsorption measures pore sizes of 0.3–300 nm, and mercury injection measures pore sizes of 3 nm–1 mm (Alonso and Alberto, 2005). Vugs and channel pores larger than a few millimetres are generally problematic to characterise because they may be larger than the plug sample. The mercury porosimeter is the only method suitable for plug-sample or

smaller sizes (Hulea and Nicholls, 2012). Therefore, use of the mercury porosimeter was deemed appropriate for the carbonate rock analysis with wide-ranging pore sizes.

2.3 Capillary pressure

For a formation with a complex pore system, capillary pressure should be determined by special core experiment. The mercury injection method is widely used because it can measure not only pore size but also capillary pressure. A non-wetting fluid such as mercury is injected into the sample, penetrating smaller pores as the injection pressure increases. The resulting measurements are expressed as a function of the cumulative penetration volume of injected mercury depending on pressure or pore size. As recommended by Purcell (1949) and Schowalter (1979), the relationship between the applied pressure and the pore throat radius into which the mercury will penetrate is given by the modified Washburn equation as follows (Daniel and Kaldi, 2010):

$$r = \frac{2IFT\cos\theta}{P_c} \tag{1}$$

where r is the pore throat radius, *IFT* is the interfacial tension of mercury and air, θ is the contact angle between mercury and the pore wall, and P_c is the applied capillary pressure.

Assuming that the pore throat sizes in the laboratory were identical to those in the reservoir, the capillary pressure for the reservoir condition ($P_{c,res}$) can be calculated as follows(Torsæter and Abtahi, 2003):

$$P_{c,res} = \frac{IFT_{res}\cos\theta_{res}}{IFT_{lab}\cos\theta_{lab}}P_{c,lab}$$
(2)

where θ is the contact angle for reservoir (*res*) and laboratory (*lab*) conditions, respectively, and $P_{c,lab}$ is the capillary pressure for the laboratory condition. Here, interfacial tension and contact angle between immiscible fluids and the reservoir rock need to be converted from laboratory conditions to reservoir conditions. However, it is difficult to determine the exact value of the contact angle for fluids in a porous matrix (Dandekar, 2013). Thus, the contact angles are often neglected and equation (2) is reformulated as follows:

$$P_{c,res} = \frac{IFT_{res}}{IFT_{lab}} P_{c,lab}$$
(3)

The IFT_{res} between mercury and air changes with pore size and the IFT_{lab} converges to 475.5 mN/m for pore sizes of 30–500 µm (Wang et al., 2016). Lombard et al. (2004) observed that all capillary pressure curves converge to a single capillary pressure curve for homogeneous samples, while the P_c data remains scattered for heterogeneous samples. Kim et al. (2011) presented superimposed type of capillary pressure model according to pore size in heterogeneous carbonate rock. Hulea and Nicholls (2012) developed a saturation model for a capillary pressure curve according to pore size in heterogeneous carbonate rock. Also, Joekar-Niasar et al. (2008) analysed the relationships among capillary pressure, interfacial area, saturation, and relative permeability in porous media. Figure 1 shows capillary pressure curves for homogeneous and heterogeneous carbonate rock, respectively.





2.4 *Relative permeability*

Determination of relative permeability is essential for reservoir productivity and multiphase flow analyses. However, this method is expensive, complex and time consuming. Roghanian et al. (2012) used correlations between residual and initial saturation to predict relative endpoint permeability and found that the results were not substantially different from experimental results. Other proposed correlations use effective phase saturation as a correlating parameter. The effective phase saturation is defined as follows (Ahmed, 2010):

$$S_o^* = \frac{S_o}{1 - S_{wc}} \tag{4}$$

$$S_w^* = \frac{S_w - S_{wc}}{1 - S_{wc}}$$
(5)

$$S_g^* = \frac{S_g}{1 - S_{wc}} \tag{6}$$

where S_o^* , S_w^* and S_g^* are the effective oil, water, and gas saturation, respectively; S_o , S_w , and S_g are the oil, water, and gas saturation, respectively; and S_{wc} is the irreducible water saturation. Several two-phase relative permeability correlations have been previously developed, including the Wyllie and Gardner, Torcaso and Wyllie, Pirson and Brooks-Corey equations. The Wyllie and Gardner equation determines relative permeability according to formation type for unconsolidated sand, cemented sandstone, and oolitic limestone (Wyllie and Gardner, 1958; Honarpour et al., 1986). The Torcaso and Wyllie equation determines relative permeability using the relative permeability for gas and the effective phase saturation; relative permeability measurements for gas are more easily captured than measurements for oil in a gas-oil system (Torcaso and Wyllie,

1958). The Pirson equation determines the wetting and non-wetting phases for water-wet rocks relative to permeability for both imbibition and drainage processes (Pirson, 1958).

The Brooks-Corey equation is most widely applied because it depicts the relationship between relative permeability and capillary pressure. Lian et al. (2015) developed a saturation model for a carbonate reservoir using the Brooks-Corey equation and recommended its use for carbonate reservoirs because it was specifically developed to describe capillary pressure curves, and effectively represents both long and short transition zones. Brooks-Corey equation modified Corey's original drainage capillary pressure–saturation relationship and combined this modified equation with Burdines's equation to develop the following expression that predicts drainage relative permeability for any pore size distribution (Corey, 1954):

$$S_w^* = \left(\frac{P_d}{P_c}\right)^\lambda \tag{7}$$

where P_d is the displacement pressure, and λ is a factor that represents pore size variation in the medium. The parameters P_c and λ are characteristic of the medium and P_d approximates the minimum drainage capillary pressure at which a continuous non-wetting phase exists in a porous medium. To determine λ and P_d , a straight trend line is produced for $(S_o - S_{oi})/(1 - S_{oi})$ versus the capillary pressure curve in a log-log scale. Using this relationship, the two-phase relative permeabilities can be determined as follows (Brooks and Corey, 1964):

$$k_{rw} = \left(S_w^*\right)^{\frac{2+3\lambda}{\lambda}} \tag{8}$$

$$k_{rmw} = \left(1 - S_w^*\right)^2 \left[1 - \left(S_w^*\right)^{\frac{2+\lambda}{\lambda}}\right] \tag{9}$$

where k_{rw} and k_{rmw} are the relative permeabilities for the wetting and non-wetting phases, respectively.

Theoretically, λ may assume any value greater than zero; larger values reflect a relatively uniform medium while smaller values reflect a medium with greater pore size variation (Honarpour et al., 1986). Belayneh et al. (2006) successfully simulated water injection using the Brooks-Corey relative permeability equation in a carbonate reservoir. Byrnes and Bhattacharya (2006) developed a capillary pressure model for a heterogeneous carbonate reservoir using a modified Brooks-Corey equation and studied the effects of initial/residual oil saturation and relative permeability on recovery. Hulea and Nicholls (2012) fitted a capillary pressure curve to a saturation model for heterogeneous carbonate rock using the Brooks-Corey equation.

In general, both micropore and macropore are developed in complex carbonate rocks. The previously developed empirical correlations described here assume a single pore system, resulting in a high degree of uncertainty in relative permeability estimation. Because capillary pressure is different according to the pore structure, precise relative permeability prediction requires consideration of pore size effects. In addition, capillary pressure and wetting characteristics must be concurrently considered to determine oil-water two-phase flow behaviour in highly heterogeneous carbonate reservoirs. The results of this study help to explain two-phase relative permeability in complex carbonate pore networks.

3 Result

3.1 Contact angle

To develop a capillary pressure relationship, the contact angle and interfacial tension between the reservoir fluids and rock must be measured. However, these measurements are too difficult to obtain, particularly for the crude oil sample. In this study, we used decane, which has an already established interfacial tension correlation with water. And, limestone core which is made for the experimented purpose is used.

When measuring the contact angle which is used as an indicator of wettability, one small drop of wetting fluid is placed on a rock surface that is already fully saturated with non-wetting fluid. Contact angle measurement (drop shape analyser – DSA100) was used to measure the contact angle. The particularly uniform LED illumination and the quality lens have high precision in order to accurately measure the contact angle or surface tension. If the contact angle of the oil droplet is greater than 90°, solids exhibit oil-wet characteristics in solid-oil-water systems (He et al., 2015).





As shown in Figure 2(a), a carbonate rock surface that does not contact oil has hydrophilic characteristics. However, a carbonate oil reservoir has contacted oil at high pressures and temperatures over thousands of years. Over this extended period, the wettability of the carbonate rock changes because of increased ionic bonds between the oil carboxyl and the carbonate rock surface. Water wetness reportedly decreases as contact time of oil increases (Zhou et al., 2000). The change of oil production rate was observed to be extremely low after aging the cores at reservoir pressure for 24 hours compared to 72 hours (Zhou et al., 1995). It means that over than 24 hours of aging time has no effect to wettability change. In this study, in order to make initial reservoir condition, carbonate core was saturated with decane for 72 hours and contact angle was measured as 98.1°. It indicates that the core used in this study shows oil-wet characteristics [Figure 2(b)].

Yang et al. (2007) reported that the contact angle increased more than 20° in an oil-wet sample as pressure increased. Ameri et al. (2013) confirmed the considerable dependence of the contact angle on pressure at the oil-wet surface. Accordingly, the wettability of reservoir rock may change to oil-wet in the reservoir condition.

3.2 End-point measurement

In order to prevent pore size change, carbonate sample was treated by pre-flushing of water at reservoir condition. Reservoir condition was assumed to be 1,100 psi and 293.15 K, respectively. After the carbonate core was fully saturated with decane using syringe pump (Figure 3), residual oil saturation and maximum water relative permeability were obtained by injecting water. Finally, irreducible water saturation and maximum oil relative permeability were measured by re-injecting oil until water was no longer flowed. Table 1 summarises these results.



Figure 3 Experimental setup for special core analysis (see online version for colours)

 Table 1
 Endpoint results from special core analysis

	Oil	Water
End point saturation	0.37	0.14
Maximum relative permeability	0.13	0.32

3.3 Mercury porosimeter

3.3.1 Pore size measurement

To estimate the relative permeability, endpoints, capillary pressure, and pore size distribution must be determined. Mercury porosimetry characterises porosity by applying various levels of pressure to a sample immersed in mercury. The pressure required for mercury to penetrate the sample's pores is inversely proportional to the size of the pores; the pore size distribution can be subsequently determined (Kantzas et al., 2014). In this study, Autopore III Porosimeter 9500 was used to measure the pore size distribution. Following transfer to a high-pressure system, the sample is surrounded by hydraulic fluid, and isostatic pressures of up to 60,000 psi are applied (Giesche, 2006; Daniel and Kaldi, 2010). Minimum pore sizes of 3.5 nm can be determined using a mercury porosimeter.

Table 2Pore system classification by pore size

Micropore	Mesopore	Macropore
<1 µm micro-pore throat diameters	1–3 μm micro-pore throat diameters	3 μm > micro-pore throat diameters

Source: Bennion and Bachu (2006)





Following mercury porosimetry analysis, the pore system is classified into three groups:

- 1 micropore
- 2 mesopore
- 3 macropore (Bennion and Bachu, 2006).

Table 2 lists the respective pore sizes in each classification, relative to a pore throat diameter. Figure 4(b) shows the experimental result of cumulative volume of injected mercury relative to pore size. If the pore size distribution has a normal distribution, the cumulative volume of injected mercury has a single bend shape. However, if the pore size is heterogeneous [see Figure 4(a)], the cumulative volume curve has a double bend shape. The carbonate sample pore system primarily comprised micropore and macropore. The amount of injected mercury varied with respect to the pore structure and injection pressure; smaller pore sizes require higher pressures for mercury penetration. In this

study, the capillary pressure and relative permeability of the carbonate with different pore sizes were analysed.





3.3.2 Capillary pressure measurement

Capillary pressure measurements captured under laboratory conditions using a mercury-air combination were converted to water-oil system that exists under reservoir condition (Brown, 1951). The pressure and temperature for the reservoir condition was assumed to be 1,100 psi and 293.15 K, respectively. To establish capillary pressure with wetting phase saturation for this reservoir condition, interfacial tension between decane and water was required. Jennings (1967) measured the interfacial tension for decane-water at pressures of 14.7–12,007 psi and temperatures of 298.15–449.15 K. Over this range of pressures and temperatures, the interfacial tension for a decane-water system can be expressed as follows:

$$IFT = 50.066 + 1.8540 \times 10^{-4} \times P - 0.12050(T - 298.15)$$
(10)

where IFT is the interfacial tension [mN/m], P is the pressure [psi] and T is the temperature [K].

A simple normalisation equation without any weighting factors or higher-order functions was used to combine mercury injection capillary pressure curves and residual oil saturation as follows (Nemes, 2016):

Relative permeability prediction

$$S_{w_n} = S_{w_n \min} + \frac{S_w - S_{w_{\min}}}{S_{w_{\max}} - S_{w_{\min}}} \times \left(S_{w_n \max} - S_{w_n \min}\right)$$
(11)

where S_w is the wetting fluid saturation of minimum (min), maximum (max) and/or in a normalised state (*n*).

In this study, mesopore were considered as macropore because of their small proportion in the sample. Capillary pressures with normalised saturation were calculated using equations (3) and (11) that pore systems were divided into macropore and micropore systems, respectively. Figure 5 depicts the superimposed capillary pressure curves. Fluid flow occurred stepwise from the macropore to the micropore. Fluids did not concurrently flow in the micropore because of the smaller displacement pressures in the macropore. Because the capillary pressure changes with the carbonate rock pore system, the effect of the pore system must be considered when explaining relative permeability in a heterogeneous carbonate.

3.4 Relative permeability prediction

To predict the relative permeability using the Brooks-Corey equation, $(S_o - S_{oi})/(1 - S_{oi})$ versus the capillary pressure curve in a log-log scale was drawn, and is depicted in Figure 6. As shown in Figure 6(a), λ and P_d cannot be determined in a heterogeneous rock because the relationship between $(S_o - S_{oi})/(1 - S_{oi})$ and capillary pressure is nonlinear. However, as shown in Figure 6(b), a strong linear trend was observed for each pore system.

Figure 6 $(S_o - S_{oi})/(1 - S_{oi})$ versus capillary pressure for the (a) combined pore system and (b) individual pore systems (see online version for colours)



If a pore system is homogeneous, P_d has a uniform value. However, our results clearly indicate large variations in P_d , suggesting a highly heterogeneous pore system in the sample. Additionally, the micropore system, with a smaller λ , had a wider range of pore sizes than the macropore system. Figure 7 depicts the capillary pressure curves with

normalised saturation that were newly calculated using λ and P_d by Brooks-Corey equation for the each pore system. P_d and residual oil saturation (S_{or}) was calculated as 9.3334 psi and 0.37 for the micropore system. In contrast, P_d and S_{or} was calculated as 0.2511 psi and 0.5 for the macropore system. It means that the resistance of micropores to water intrusion is much higher than the case of macropores.

Figure 7 Capillary pressure curves derived according to the pore system (see online version for colours)



Next, the relative permeability was calculated using the Brooks-Corey equation for the pore system shown in Figure 8. As shown in Figure 8(a), a small reduction in oil saturation drastically reduced the relative permeability of oil. For the macropore system, the relative permeability of oil is almost zero below an oil saturation of 0.9. Under these conditions, only water that occupies the macropore flows. As shown in the water relative permeability curve, water begins to flow at relatively high oil saturation and water penetrates more easily in the macropores because of a smaller P_d . Thus, in oil-wet condition for the macropore system, oil is stationary and water flows easily through the central pore network at high oil saturations. In practice, when water flooding is applied to a carbonate reservoir, water cannot effectively displace oil, and early breakthroughs occur due to the preferential flow of water in the macropore system.

In contrast, as shown in Figure 8(b), the intersection of relative permeabilities for water and oil in the micropore system occurred at an oil saturation of 0.7 indicating an oil-wet condition. This means that water flows better as oil saturation decreases, and substantial amount of oil is trapped due to the high residual oil saturation. However, greater pressures are required for the water to penetrate the pore; oil preferentially occupies the smaller pore spaces where capillary forces are the greatest. Therefore, in practice, it is very difficult to enter the water into micropore in heterogeneous carbonate reservoir because most water flows through the macropore.





4 Conclusions

This study presents a relative permeability estimation method with respect to the heterogeneous pore network that considers the capillary pressure, contact angle, pore size distribution, and residual oil saturation. As a result, the pore size distribution index of the macropore system was greater than that of the micropore system, indicating a wider pore size range for the micropore system. The results of this study regarding capillary pressure based on λ and P_d suggest that the relative permeability of the carbonate sample was likely affected by pore size. Therefore, oil and water were observed to flow differently through micropore and macropore, according to the oil saturation and stable water displacement was not assured, leading to an early breakthrough. In conclusion, the preferential flow of water through the macropore and the resistance of micropore to water penetration are the primary reasons for low oil recovery in carbonate reservoirs. Thus, improved water penetration for the micropore system which can be accomplished by increasing water injection pressure or water-oil capillary pressure is the key to enhanced oil recovery. However, for the micropore system, increasing capillary pressure is difficult to achieve because water flows through the highly permeable macropore. For this reason, capillary pressure cannot be maintained above the pressure level required for water penetration, resulting in a very small amount of recoverable oil in the micropore system.

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